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PTO-1590 (8-01)

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FILE 'LREGISTRY' ENTERED AT 15:19:28 ON 07 JUL 2005
L1
               STR
     FILE 'REGISTRY' ENTERED AT 15:31:31 ON 07 JUL 2005
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     FILE 'CAOLD' ENTERED AT 15:38:28 ON 07 JUL 2005
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     FILE 'ZCA' ENTERED AT 15:39:08 ON 07 JUL 2005
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        32218 S L4 OR TRIETHANOLAMINE# OR TRIETHANOL#(A)AMINE# OR TRI(A
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           482 S (L9 OR L10) AND L11
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           1888 S L4 (L) RACT/RL
L13
           3321 S L5 (L) RACT/RL
L14
          5803 S L6 (L) RACT/RL
L15
             1 S L13 AND L14 AND L15
L16
         33722 S SHALE#
L17
         213426 S DRILL? OR BORE# OR BORING# OR BOREING# OR FRACTUR?
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         14797 S OILWELL? OR OIL#(2A)(WELL OR WELLS) OR DERRICK?
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FILE 'ZCA' ENTERED AT 17:03:43 ON 07 JUL 2005
L22 822634 S L20 OR L21 OR HYDROGENA? OR RANEY#
L23 48 S L12 AND L22
L24 0 S L23 AND (L17 OR L18 OR L19)
L25 1 S L12 AND (L17 OR L18 OR L19)
L26 14 S L8 OR L16 OR L25
L27 48 S L23 NOT L26

FILE 'REGISTRY' ENTERED AT 17:11:51 ON 07 JUL 2005

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L1 STR

2 4 6 8 10 12 H2N CH2 CH2 CH2 CH2 G2 O\CH2 CH2 CH2 NH2 CH2 CH2 CH2 G1 CH2 @37 13 14 15 16 3 5 7 9 11

 $N \sim CH2 \cdot CH2 \cdot OH$ $N \sim CH2 \cdot CH2 \cdot O \sim CH2 \cdot CH2 \cdot CH2 \cdot NH2$ @38 19 20 21 @39 24 25 26 27 28 29 30

N~CH2-CH2-CH2-NH2 @40 33 34 35 36

VAR G1=NH/38/39/40 VAR G2=OH/37 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 33

STEREO ATTRIBUTES: NONE

L3 6 SEA FILE=REGISTRY SSS FUL L1

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6 ANSWERS

FILE 'ZCA' ENTERED AT 17:12:26 ON 07 JUL 2005 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2005 AMERICAN CHEMICAL SOCIETY (ACS)

=> d 126 1-14 cbib abs hitstr hitind

L26 ANSWER 1 OF 14 ZCA COPYRIGHT 2005 ACS on STN

142:282555 Shale hydration inhibition agent and method of use. Patel, Arvind D.; Stamatakis, Emanuel; Davis, Eric; Friedheim, Jim (M I LLC, USA). U.S. Pat. Appl. Publ. US 2005049150 A1 20050303, 15 pp. (English). CODEN: USXXCO. APPLICATION: US 2003-647830 20030825.

A water-base fluid for use in drilling, cementing, workover, AΒ fracturing and abandonment of subterranean wells through a formation contg. a shale which swells in the presence of water. In one illustrative embodiment, the drilling fluid includes, an aq. based continuous phase, a weighting agent, and a shale hydration inhibition agent. The shale hydration inhibition agent should have the general formula: 1 in which A is independently selected from H and CH 2 CH 2 CH 2 NH 2; and in which B is independently selected from H, CH 2 CH 2 OH, CH 2 CH 2 OCH 2 CH 2 CH 2 NH 2 and CH 2 CH 2 CH 2 NH 2. The shale hydration inhibition agent is present in sufficient concn. to substantially reduce the swelling of shale drilling cuttings upon contact with the drilling fluid. drilling fluid may be formulated to include a wide variety of components of aq. based drilling fluids, such as weighting agents, fluid loss control agents, suspending agents, viscosifying agents, rheol. control agents, as well as other compds. and materials known to one of skill in the art. The fluids may also be used in the disposal by reinjection of drilling cutting into a selected subterranean disposal formation.

IT 61579-07-5, RMR 13-21A 61579-10-0 73947-23-6 847238-40-8 847257-78-7, RMR 13-21B

(shale hydration inhibition agent and method of use)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

 $\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{O-} \text{(CH}_2)_3-\text{NH}_2 \\ | \\ \text{H}_2\text{N--} \text{(CH}_2)_3-\text{O-} \text{CH}_2-\text{CH}_2-\text{N--} \text{CH}_2-\text{CH}_2-\text{O--} \text{(CH}_2)_3-\text{NH}_2 \end{array}$

RN 61579-10-0 ZCA CN Ethanol, 2-[bis[2-(3-aminopropoxy)ethyl]amino]- (9CI) (CA INDEX NAME)

CH2-CH2-OH $H_2N - (CH_2)_3 - O - CH_2 - CH_2 - N - CH_2 - CH_2 - O - (CH_2)_3 - NH_2$ 73947-23-6 ZCA RN Ethanol, 2-[[2-(3-aminopropoxy)ethyl]amino]- (9CI) (CA INDEX NAME) CN $HO-CH_2-CH_2-NH-CH_2-CH_2-O-(CH_2)_3-NH_2$ RN847238-40-8 ZCA Ethanol, 2,2'-[[2-(3-aminopropoxy)ethyl]imino]bis- (9CI) (CA INDEX CN CH2-CH2-OH $HO-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$ RN 847257-78-7 ZCA 1-Propanamine, 3,3'-[iminobis(2,1-ethanediyloxy)]bis-(9CI) (CA CN INDEX NAME) $H_2N-(CH_2)_3-O-CH_2-CH_2-NH-CH_2-CH_2-O-(CH_2)_3-NH_2$ IC ICM C09K007-02 INCL 507136000 51-2 (Fossil Fuels, Derivatives, and Related Products) CC 102-71-6D, Triethanolamine, mono, di, and triethers with IT 111-42-2D, Diethanolamine, reaction products propanolamine, uses with acrylonitrile, hydrogenated 471-34-1, Calcium carbonate, uses 1317-60-8, Hematite, uses 1332-37-2, Iron oxide, uses 11138-66-2, DuoVis 13397-26-7, Calcite (Ca(CO3)), uses 13462-86-7, Barite 61579-07-5, RMR 13-21A 130392-39-1, Rev-Dust 61579-10-0 73947-23-6 847256-75-1, UltraFree 847238-40-8 847257-76-5, PolyPac **847257-78-7**, RMR 13-21B (shale hydration inhibition agent and method of use) L26 ANSWER 2 OF 14 ZCA COPYRIGHT 2005 ACS on STN 103:115382 Chemiluminescence vs. Kjeldahl determination of nitrogen in

oil shale retort waters and organonitrogen compounds.

57(12), 2320-5 (English) 1985. CODEN: ANCHAM.

Jones, Bonnie M.; Daughton, Christian G. (Lawrence Berkeley Lab., Univ. California, Berkeley, CA, 94720, USA). Analytical Chemistry,

ISSN: 0003-2700.

The applicability of combustion/chemiluminescent N detn. to detg. N AB in oil shale wastewaters and various representative chem. classes was demonstrated. Only azoxy compds. and those contg. the pyrazole nucleus were not amenable to anal. The majority of 56 compds. tested yielded 90-110% of their theor. N content; enhanced recovery was found for N oxide salts. For 12 oil shale wastewaters, combustion/chemiluminescence gave total N values (1100-28,800 mg/L) that did not differ statistically (P > 0.10) from those obtained by the time-consuming wet-chem. Kjeldahl method. relative std. deviations for 10 replicates of each wastewater were less than 3.5%. No matrix or solvent effects were found. **75-05-8**, analysis **111-42-2**, analysis IT(nitrogen detn. in, by combustion and chemiluminescence) 75-05-8 ZCA RN . Acetonitrile (8CI, 9CI) (CA INDEX NAME) CN $H3C-C \equiv N$ RN 111-42-2 ZCA Ethanol, 2,2'-iminobis- (9CI) (CA INDEX NAME) CN HO-CH2-CH2-NH-CH2-CH2-OH CC 80-6 (Organic Analytical Chemistry) Section cross-reference(s): 61 nitrogen detn org chemiluminescence; wastewater analysis nitrogen ST chemiluminescence; oil shale wastewater analysis nitrogen IT(pyrolysis of, nitrogen detn. in retort wastewaters from, comparison of chemiluminescence and Kjeldahl methods for) 7727-37-9, analysis ΙT (detn. of, in oil shale retort wastewaters and org. compds., comparison of chemiluminescence and Kjeldahl methods for) IT 7732-18-5, analysis (nitrogen detn. in waste-, from oil shale retorts, comparison of chemiluminescence and Kjeldahl methods for) 51-17-2 56-40-6, analysis 57-13-6, analysis 59-67-6, analysis IT 60-00-4, analysis 68-12-2, analysis **75-05-8**, 67-51-6 83-07-8 86-74-8 88-75**-**5 91-22-5, analysis 96-54-8 analysis 108-47-4 100-02-7, analysis 100-71-0 104-90-5 108-48-5 108-89-4 108-99-6 109-00-2 109-05-7 108-75-8 108-80-5 109-08-0 110-85-0, analysis 110-86-1, analysis 109-06-8 110-89-4, analysis **111-42-2**, analysis 119-65-3 120-72-9, analysis 142-08-5 271-44-3 288-13-1 288-32-4, 536-78-7 536-75-4 analysis 289-80-5 290-37-9 529-21-5

554-84-7 613-50-3 622-39-9 872-50-4, analysis 931-20-4 934-48-5 1462-84-6 1484-12-4 1562-94-3 3279-76-3 5144-11-6 7757-79-1, analysis 7758-09-0 20815-29-6 27464-82-0 56430-08-1

(nitrogen detn. in, by combustion and chemiluminescence)

- L26 ANSWER 3 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 100:142007 Metallic dispersant-detergent additive of high alkalinity for lubricating oils. Le Coent, Jean Louis; Demoures, Bernard (Societe Orogil, Fr.). Fr. Demande FR 2529225 A1 19831230, 14 pp. (French). CODEN: FRXXBL. APPLICATION: FR 1982-11059 19820624.
- The prepn. of the title compds. involved the following: (1) carbonation of Mg alkanesulfonate-sulfurized Ca alkylphenoxide-MgO-water-glycol-amine-oil mixt., (2) removal of glycol and water, (3) filtration. Thus, an effective, oil-sol. dispersant-detergent was prepd. as above starting with sulfonic acid, mol. wt. 470, and dodelcylphenol; the amine was either ethylenediamine [107-15-3] or tris(6-amino-3-oxahexyl)amine [61579-07-5].
- IT **61579-07-5**

(adjuncts, in carbonation of overbased calcium alkylphenoxides and magnesium alkylsulfonates)

- RN 61579-07-5 ZCA
- CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{O-} \text{(CH}_2)_3-\text{NH}_2 \\ | \\ \text{H}_2\text{N--} \text{(CH}_2)_3-\text{O-} \text{CH}_2-\text{CH}_2-\text{N--} \text{CH}_2-\text{CH}_2-\text{O--} \text{(CH}_2)_3-\text{NH}_2 \end{array}$$

- IC C10M001-38
- CC 51-8 (Fossil Fuels, Derivatives, and Related Products)
- IT **61579-07-5**

(adjuncts, in carbonation of overbased calcium alkylphenoxides and magnesium alkylsulfonates)

- L26 ANSWER 4 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 100:142006 Overbased detergent-dispersant additive for lubricating oils. Le Coent, Jean Louis; Demoures, Bernard (Societe Orogil, Fr.). Fr. Demande FR 2529224 A1 19831230, 14 pp. (French). CODEN: FRXXBL. APPLICATION: FR 1982-11058 19820624.
- The prepn. of carbonated-sulfurized Ca alkylphenoxides was carried out by the following steps: (1) carbonation of Mg alkylbenzenesulfonate-sulfurized alkylphenol-MgO-glycol-amine mixt. in a diluent oil (2) neutralization with lime, (3) 2nd carbonation, (4) solvent removal by distn., and (5) filtration. Thus, dodecylphenol [27193-86-8] was converted to sulfurized Mg dodecylphenoxide and treated as outlined above (the amine was

ethylenediamine [107-15-3]) to give an effective, oil-sol. detergent-dispersant for lubricating oils.

IT **61579-07-5**

(adjuncts, for prepn. of carbonated-sulfurized overbased phenoxides)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{O-}\left(\text{CH}_2\right)_3-\text{NH}_2 \\ | \\ \text{H}_2\text{N-}\left(\text{CH}_2\right)_3-\text{O-}\left(\text{CH}_2-\text{CH}_2-\text{N-}\left(\text{CH}_2-\text{CH}_2-\text{O-}\left(\text{CH}_2\right)_3-\text{NH}_2\right)\right. \end{array}$$

Ι

IC C10M001-38

CC 51-8 (Fossil Fuels, Derivatives, and Related Products)

L26 ANSWER 5 OF 14 ZCA COPYRIGHT 2005 ACS on STN
98:180416 Epoxy resin compositions. Kluger, Edward W. (Milliken
Research Corp., USA). U.S. US 4352920 A 19821005, 5 pp.
(English). CODEN: USXXAM. APPLICATION: US 1981-230719 19810202.
GI

 $\begin{array}{c} \text{R}^5 \quad \text{R}^3 \\ \mid \quad \mid \quad \\ \text{CH}_2\text{CHOCHCH}_2\text{CH}_2\text{NH}_2 \\ \text{R}^1 - \text{N} \\ \text{CH}_2\text{CHOR}^2 \\ \mid \quad \\ \text{R}^4 \end{array}$

Polyamines (I; R',R2 = H, C(R3)HCH2CH2NH2; R3, R4, R5 = C1-6 alkyl) prepd. by cyanoalkylation of dialkanolamines and redn. of the resulting aminonitriles are effective curing agents for epoxy resins in compns. contg. 15-50 parts I per 100 parts resin. Thus, bis(2-cyanoethoxy)-N-(2-cyanoethyl)diethanolamine [85438-88-6] was reduced with H over Ru or alumina in the presence of NH3 at 125-130.degree./1600-1700 psi to give bis(2-aminopropoxy)-N-(2-aminopropyl)diethanolamine [76461-16-0], 21.4 parts of which were combined with 10 parts epoxy resin based on bisphenol A diglycidyl ether, poured into a mold and cured 2 h at 80.degree. and 2 h at 150.degree. The crosslinked product had glass transition

temp. 100.3.degree..

IT 76461-16-0

(crosslinking agents, for epoxy resins)

RN 76461-16-0 ZCA

CN 1,3-Propanediamine, N,N-bis[2-(3-aminopropoxy)ethyl]- (9CI) (CA INDEX NAME)

$$CH_2 - CH_2 - O - (CH_2)_3 - NH_2$$

 $H_2N-(CH_2)_3-O-CH_2-CH_2-N-(CH_2)_3-NH_2$

IC C08G059-50; C08G059-64

INCL 528111000

CC 37-6 (Plastics Manufacture and Processing)

IT 76461-16-0

(crosslinking agents, for epoxy resins)

- L26 ANSWER 6 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 95:170776 Improving the compatibility of plasticizers and fillers in polymers. Machurat, Jean; Morawski, Jean Claude; Soula, Gerard (Rhone-Poulenc Industries S. A., Fr.). Eur. Pat. Appl. EP 32076 19810715, 36 pp. (French). CODEN: EPXXDW. APPLICATION: EP 1980-401751 19801208.
- Alkenylsuccinimides [e.g., polyisobutenyl- or AΒ tetrapropenylsuccinimide (I)], prepd. by the reaction of alkenylsuccinic anhydrides with [H2N(CH2)3OCH2CH2]3N, [H2N(CH2)3OCH2]2, (H2NCH2)2, p-(H2N)2C6H4, [H2N(CH2)6]2NH, or H2N(CH2CH2NH)4H (II), improve the compatibility of hydrocarbon oils, [117-81-7], and other plasticizers with fillers such as SiO2, Na aluminosilicate, kaolin, or chalk in the prepn. of filled, plasticized rubbers. Thus, I was prepd. by heating 665 g tetrapropenylsuccinic anhydride to 130.degree. in the presence of Br, adding 189 g II during 30 min, and heating the mixt. to 160.degree./25 mm. A mixt. of 100 g SiO2 and .apprx.400 mL arom. hydrocarbon oil (Dutrex 729 FC) contg. 1% I flowed on a surface angled at 45.degree.. Without I, .apprx.600 mL oil was required to give a flowable mixt. The addn. of 0.6 part I to a compn. contg. SBR 60, butadiene rubber 40, SiO2 60, and Dutrex 729 FC 20 parts decreased the viscosity of the compn. and increased the modulus and decreased the permanent deformation of vulcanizates.
- IT 61579-07-5D, reaction products with alkenylsuccinic anhydrides

(plasticizers contg., for compatibility of fillers with rubber compns.)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{O-} \text{ (CH}_2)\text{ }_3-\text{NH}_2\\ | & \cdot \\ \text{H}_2\text{N--} \text{ (CH}_2)\text{ }_3-\text{O--} \text{CH}_2-\text{CH}_2-\text{N--} \text{CH}_2-\text{CH}_2-\text{O--} \text{ (CH}_2)\text{ }_3-\text{NH}_2 \end{array}$$

IC C08L021-00; C08K005-00

CC 38-9 (Elastomers, Including Natural Rubber)

106-50-3D, reaction products with alkenylsuccinic anhydrides 107-15-3D, reaction products with alkenylsuccinic anhydrides 108-30-5D, alkenyl derivs., reaction products with polyamines 112-57-2D, reaction products with alkenylsuccinic anhydrides 123-56-8D, alkenyl and amino derivs. 143-23-7D, reaction products with alkenylsuccinic anhydrides 2997-01-5D, reaction products with alkenylsuccinic anhydrides 61579-07-5D, reaction products with alkenylsuccinic anhydrides

(plasticizers contg., for compatibility of fillers with rubber compns.)

L26 ANSWER 7 OF 14 ZCA COPYRIGHT 2005 ACS on STN

- 94:66480 Light-stable polyurethanes for integral foams and elastomers. Horacek, Heinrich; Volkert, Otto (Kunststofflab., Badischen Anilin und Sodafabrik A.-G., Ludwigshafen am Rhein, D-6700, Fed. Rep. Ger.). Angewandte Makromolekulare Chemie, 90, 109-42 (German) 1980. CODEN: ANMCBO. ISSN: 0003-3146.
- The photodegrdn. of polyurethanes prepd. from arom. polyisocyanates can be prevented by coating or lining them with films, but these methods are expensive. The use of lightfast, aliph. polyisocyanates gives trouble, particularly in reaction-injection molding, owing to their relatively low reactivity. Polyamide catalysts were developed which increased the rate of polymn. of aliph. polyisocyanates to the required levels, enabling the prepn. of polyurethanes equiv. in properties to those from arom. polyisocyanates, and with satisfactory lightfastness. These polymers are potentially useful in flexible and rigid integral-skin foam, reaction-injection molding of elastomers for the automotive industry, and in-mold coating.

IT **76461-16-0**

(catalysts, for polyurethane prepn. from aliph. polyisocyanates)

RN 76461-16-0 ZCA

CN 1,3-Propanediamine, N,N-bis[2-(3-aminopropoxy)ethyl]- (9CI) (CA INDEX NAME)

$$CH_2-CH_2-O-(CH_2)_3-NH_2$$

 $H_2N-(CH_2)_3-O-CH_2-CH_2-N-(CH_2)_3-NH_2$

CC 36-3 (Plastics Manufacture and Processing)

IT 6864-37-5 9002-98-6 39423-51-3 72088-96-1 76461-15-9 **76461-16-0**

(catalysts, for polyurethane prepn. from aliph. polyisocyanates)

L26 ANSWER 8 OF 14 ZCA COPYRIGHT 2005 ACS on STN

- 94:35075 Action of nitrogen compounds on aluminum. Yoshimura, Chozo; Ogura, Toshiaki (Coll. Sci. Eng., Kinki Univ., Osaka, Japan). Aruminyumu Kenkyu Kaishi, 146, 55-6 (Japanese) 1980. CODEN: AKKADN. ISSN: 0285-5224.
- The wt. change and elec. potential change of Al in HCONMe2 [68-12-2], pyridine [110-86-1], ethanolamine [141-43-5], diethanolamine [111-42-2], triethanolamine [102-71-6], aniline [62-53-3], nitrobenzene [98-95-3], and acetontrile [75-05-8] were studied. Generally, the action of amphoteric solvents was stronger than that of nonprotonic solvents.
- IT 75-05-8, reactions 102-71-6, reactions
 111-42-2, reactions

(corrosion by, of aluminum)

RN 75-05-8 ZCA

CN Acetonitrile (8CI, 9CI) (CA INDEX NAME)

H3C-C=N

RN 102-71-6 ZCA

CN Ethanol, 2,2',2''-nitrilotris- (9CI) (CA INDEX NAME)

 $\rm CH_2-CH_2-OH$

 $HO-CH_2-CH_2-N-CH_2-CH_2-OH$

RN 111-42-2 ZCA

CN Ethanol, 2,2'-iminobis- (9CI) (CA INDEX NAME)

HO-CH2-CH2-NH-CH2-CH2-OH

- CC 56-8 (Nonferrous Metals and Alloys)
- L26 ANSWER 9 OF 14 ZCA COPYRIGHT 2005 ACS on STN 94:30184 .gamma.-Aminopropoxy compounds. Polievka, Milan; Balak, Jiri; Macho, Vendelin (Czech.). Czech. CS 181846 19800215, 6 pp. (Slovak). CODEN: CZXXA9. APPLICATION: CS 1974-1226 19740220.

AB The title compds. were prepd. by cyanoethylation of polyalcs. and catalytic hydrogenation of the products. Thus, a mixt. of 590 g polyethylene glycol (mol. wt. 600), 600 mL water, and 2 g NaOH was treated at 30.degree. with 2 mols CH2:CHCN 4 h and the product was hydrogenated over Raney Ni to yield 680 g of wax-like .alpha.-w-diaminopolyethylene glycol (mol. wt. 693).

IT 61579-07-5P

(prepn. of)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

IC C07C091-02

CC 23-9 (Aliphatic Compounds)

Section cross-reference(s): 35

IT 5045-94-3P 16499-88-0P 16728-59-9P 24991-53-5P 62035-48-7P 76126-99-3P

(prepn. of)

- L26 ANSWER 10 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 93:28184 Surfactants containing fluoroalkyl groups. Hayashi, Takao; Ohtoshi, Yukio (Asahi Glass Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 55007818 19800121 Showa, 7 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 1978-78720 19780630.
- AB RCONR1ZOZ1NR2Z2CO2R3 (R = C1-20 fluoroalkyl; R1 = H, alkyl; R2 = alkyl, hydroxyalkyl; R3 = H, NH4, alkali metal, Z,Z1 = C1-10 alkylene; Z2 = C1-5 alkylene) were prepd., had low crit. micelle concn. and showed high surface activity at low concn. Thus, C9F19CO2CHMe2 (I) [71356-37-1] in Me2CHOH was treated with H2N(CH2)3OCH2CH2NHCH2CH2OH [73947-23-6] at 70.degree. for 20 h to give >98% C9F19CONH(CH2)3OCH2CH2NHCH2CH2CO2H [73947-24-7] which was then treated with C1CH2CO2H [79-11-8] in Me2CHOH to give C9F19CONH(CH2)3OCH2CH2N(CH2CHOH)CH2CO2H (II) [73947-25-8] in 85% yield based on I. A 0.1(0.001)% aq. II had surface tension 16.4(18.4) dynes/cm, compared with 17.0(31.7) for C9F19CONH(CH2)3N+Me2CH2CO2-.

IT **73947-23-6**

(reaction of, with perfluorodecanoic acid ester)

RN 73947-23-6 ZCA

CN Ethanol, 2-[[2-(3-aminopropoxy)ethyl]amino]- (9CI) (CA INDEX NAME)

 $HO-CH_2-CH_2-NH-CH_2-CH_2-O-(CH_2)_3-NH_2$

- IC C11D001-10; C07C103-38
- CC 46-3 (Surface Active Agents and Detergents)
- IT 73947-23-6

(reaction of, with perfluorodecanoic acid ester)

- L26 ANSWER 11 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 91:125917 Alkenylamines useful as additives for lubricating oils and fuels. Soula, Gerard (Orogil S. A., Fr.). Fr. Demande FR 2381067 19780915, 14 pp. (French). CODEN: FRXXBL. APPLICATION: FR 1977-5006 19770222.
- Fuel and lubricating oils having detergent and dispersant properties AB contain .apprx.50 mmol alkenylamine/kg oil. The alkenylamine is prepd. by treating (HOCHR1CH) nN (CH2CHR2CH2NH2) 3-(m+n) (CH2CHR1OCH2CHR2CH2NH2)n (R1 and R2 = H or Me; C1-4 alkyl, or Ph; m = 1-3; n = 0-2) with a brominated or chlorinated polyisobutylene (mol. wt. 600-4200; 1-1.5 g-atom halogen/mol halogenated polyisobutylene) at 130-80.degree. and amine-halogenated polyisobutylene mol ratio 0.25-2:1. Thus, an amine was prepd. by heating brominated polyisobutylene (1 mol; 2.5 wt.% Br; mol. wt. 3400), tris(6-amino-3-oxahexyl)amine [61579-07-5] (1.8 mol), and octanol to 160.degree. for 15 h, cooling to 90.degree., adding 10% aq. NaOH, and refluxing the mixt. for 1 h. SAE 30 oil (20 g; contg. 5 g engine sludge including 2% carbonaceous material) was tested for antifoam, antirust, and dispersant properties in the presence of the amine, Ca alkylbenzenesulfonate, overbased Ca alkylphenate, and Zn dihexyl phosphorodithioate [7282-28-2].
- IT **61579-07-5D**, reaction products with halogenated polyisobutylene

(detergents-dispersants, for fuels and lubricating oils)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$_{\text{L}}^{\text{CH}_2-\text{CH}_2-\text{O}-\text{(CH}_2)}_{3}-\text{NH}_2$$

 $H_2N-(CH_2)_3-O-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$

- IC C08F008-32; C08F008-18; C08F010-00; C10L001-22
- CC 51-7 (Fossil Fuels, Derivatives, and Related Products)
 Section cross-reference(s): 37
- 9003-27-4D, halogenated, reaction products with amines **61579-07-5D**, reaction products with halogenated polyisobutylene

(detergents-dispersants, for fuels and lubricating oils)

L26 ANSWER 12 OF 14 ZCA COPYRIGHT 2005 ACS on STN

- 90:25861 Additive for lubricants and fuels. Soula, Gerard (Societe Orogil, Fr.). Ger. Offen. DE 2806908 19780824, 17 pp. (German). CODEN: GWXXBX. APPLICATION: DE 1978-2806908 19780217.
- AB High-mol.-wt. alkenylamines, prepd. by treating a polyamine (a hydrogenated cyanoethylated alkanolamine) and a halogenated polyolefin were detergent-dispersant additives with anticorrosion and antisludge properties. Thus, 0.2 mol chlorinated poly(isobutylene) and 0.67 mol tris(6-amino-3-oxahexyl)amine reacted in the presence of Na2CO3 to form a viscous product with 1.5% N and 0.5% residual Cl. When tested at 50 mmol/kg SAE 30 oil it showed better dispersion, antirust, and antisludge properties than a com. additive prepd. from tetraethylenepentamine and chlorinated poly(isobutylene).
- IT 61579-07-5D, polyisobutenyl derivs. (lubricating oil detergents)
- RN 61579-07-5 ZCA
- CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} & \text{CH}_2-\text{CH}_2-\text{O-} \text{(CH}_2)_3-\text{NH}_2 \\ & | & \\ \text{H}_2\text{N-} \text{(CH}_2)_3-\text{O-} \text{CH}_2-\text{CH}_2-\text{N-} \text{CH}_2-\text{CH}_2-\text{O-} \text{(CH}_2)_3-\text{NH}_2 \\ \end{array}$$

- IC C07C093-02
- CC 51-7 (Fossil Fuels, Derivatives, and Related Products)
- IT 61579-07-5D, polyisobutenyl derivs. (lubricating oil detergents)
- L26 ANSWER 13 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 87:104344 Lubricant additives comprising alkenylsuccinic acid imides. Soula, Gerard; Duteurtre, Philippe (Societe Orogil, Fr.). Ger. Offen. DE 2616751 19761028, 32 pp. (German). CODEN: GWXXBX. APPLICATION: DE 1976-2616751 19760415.
- Alkenylsuccinimide derivs. of ether-contg. polyamines are useful as lubricant additives. Thus, adding 58 g N(CH2CH2OCH2CH2CH2NH2)3 [61579-07-5] (prepd. by cyanoethylation of triethanolamine [102-71-6] and hydrogenation) over 1 h to 900 g (polyisobutenyl)succinic anhydride (acid no. 74 mg KOH/g, prepd. from polyisobutylene with mol. wt. 1000) stirred at 120.degree. and stirring 3 h at 160.degree. and 50 mm gives a clear triimide (I) contg. 1.06% N. Motor testing of SAE 30 motor oil contg. I 50, Ca alkylbenzenesulfonate 30, overbased Ca alkylphenoxide 30, and Zn dihexyl dithiophosphate [7282-28-2] 15 mmol/kg gives dispersion rating 470, corrosion protection rating 16, and antifoam rating 10-05, compared with 400, 12, and 580-450, resp., with (polyisobutenyl)succinimide from triethylenetetramine in place of I.
- IT 61579-10-0P

(manuf. and reaction of with polyisobutenylsuccinic anhydride)

RN 61579-10-0 ZCA

CN Ethanol, 2-[bis[2-(3-aminopropoxy)ethyl]amino]- (9CI) (CA INDEX NAME)

 $H_2N-(CH_2)_3-O-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$

IT **61579-07-5P**

(manuf. and reaction of, with polyisobutenylsuccinic anhydride)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$CH_2 - CH_2 - O - (CH_2)_3 - NH_2$$

 $H_2N-(CH_2)_3-O-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$

IC C10M001-36

CC 51-7 (Fossil Fuels, Derivatives, and Related Products)
Section cross-reference(s): 27

IT 61579-10-0P

(manuf. and reaction of with polyisobutenylsuccinic anhydride)

IT **61579-07-5P** 61579-09-7P 61642-85-1P

(manuf. and reaction of, with polyisobutenylsuccinic anhydride)

- L26 ANSWER 14 OF 14 ZCA COPYRIGHT 2005 ACS on STN
- 86:43167 Polyamines containing ether groups. Collet, Paul (Rhone-Poulenc S. A., Fr.). Ger. Offen. DE 2616750 19761028, 18 pp. (German). CODEN: GWXXBX. APPLICATION: DE 1976-2616750 19760415.

AB H2N(CH2)3O(CH2)2NRR1 [I; R = Et, H2N(CH2)3O(CH2)2, HOCH2CH2; R1 = H2N(CH2)3O(CH2)2, HOCH2CH2] were prepd. by the cyanoethylation of a hydroxyethyl amine, followed by redn. of the intermediate nitrile. Thus, (HOCH2CH2)3N reacted with acrylonitrile in NaOH at 36.degree. to give N(CH2CH2OCH2CH2CN)3, which was hydrogenated over Raney Ni to I [R = R1 = H2N(CH2)3O(CH2)2].

IT 61579-07-5P 61579-10-0P

(prepn. of)

RN 61579-07-5 ZCA

CN 1-Propanamine, 3,3',3''-[nitrilotris(2,1-ethanediyloxy)]tris- (9CI) (CA INDEX NAME)

$$H_2N-(CH_2)_3-O-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$$

RN 61579-10-0 ZCA

CN Ethanol, 2-[bis[2-(3-aminopropoxy)ethyl]amino]- (9CI) (CA INDEX NAME)

$$H_2N-(CH_2)_3-O-CH_2-CH_2-N-CH_2-CH_2-O-(CH_2)_3-NH_2$$

- IC C07C093-04
- CC 23-4 (Aliphatic Compounds)
- IT **61579-07-5P** 61579-09-7P **61579-10-0P** 61642-85-1P (prepn. of)
- => d 127 1-48 ti
- L27 ANSWER 1 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Transition metal-carbide and nitride containing catalysts , their preparation, and use as oxidation and dehydrogenation catalysts
- L27 ANSWER 2 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI An investigation of alternative catalytic approaches for the direct synthesis of hydrogen peroxide from hydrogen and oxygen
- L27 ANSWER 3 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Populations at risk
- L27 ANSWER 4 OF 48 ZCA COPYRIGHT 2005 ACS, on STN
- TI Preparation of aminophenol derivatives and their use in coloring agents
- L27 ANSWER 5 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Ferrous Ions as Catalysts for Photochemical Reduction of CO2 in Homogeneous Solutions
- L27 ANSWER 6 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Combustion chemistry of HAN, TEAN, and XM46
- L27 ANSWER 7 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Increased cytotoxic sensitivity of cultured FHM fish cells by

RICHARD 10/647,830

simultaneous treatment with sodium dodecyl sulfate and buthionine sulfoximine

- L27 ANSWER 8 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Fate coefficients for the toxicity assessment of air pollutants
- L27 ANSWER 9 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Crystalline metal-organic microporous materials
- L27 ANSWER 10 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Salt-blend dissociation for manufacture of coated hard powders for sintered carbide alloys or metal composite materials
- L27 ANSWER 11 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Methodical analysis of protein-nitrocellulose interactions to design a refined digestion protocol
- L27 ANSWER 12 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Semiconductor photocatalysis. ZnS-nanocrystallite-catalyzed photooxidation of organic compounds
- L27 ANSWER 13 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Federal facility compliance demonstration with state air toxics regulations
- L27 ANSWER 14 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI French limiting values for occupational exposure to chemicals
- L27 ANSWER 15 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Effect of the cationic surfactants on the spectrophysical and catalytic characteristics of tetrabromothiofluorescein
- L27 ANSWER 16 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Process for removal of trace polar contaminants from light olefin streams
- L27 ANSWER 17 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Vanadium(V)-protein model studies: solid-state and solution structure
- L27 ANSWER 18 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Efficient and selective electron mediation of cobalt complexes with cyclam and related macrocycles in the p-terphenyl-catalyzed photoreduction of carbon dioxide
- L27 ANSWER 19 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Air contaminants

- L27 ANSWER 20 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI 4,4'-Bis(dimethylamino)benzophenone (Michler's ketone) a common indicator for the determination of the acidity and dipolarity/polarizability of reaction media
- L27 ANSWER 21 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- process for the preparation of [bis[2-(alkylcarbamoyl)ethyl]amino] acetonitrile derivatives and their use for the preparation of antiarrhythmic N,N-bis[2-[(alkylcarbamoyl)oxy]ethyl]-1,2-ethanediamines
- L27 ANSWER 22 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Enhanced p-terphenyl-catalyzed photoreduction of carbon dioxide to carbon monoxide through the mediation of a cobalt(III)-cyclam complex
- L27 ANSWER 23 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Correlation of the neutral red uptake inhibition assay of cultured fathead minnow fish cells with fish lethality tests
- L27 ANSWER 24 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Kinetic and electron paramagnetic resonance studies of photochemical reactions of gas mixtures of methane, ammonia and water
- L27 ANSWER 25 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Kinetic and electron paramagnetic resonance studies of photochemical reactions of gas mixtures of methane, ammonia and water
- L27 ANSWER 26 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Remarkable solvent effects on the photocatalytic behavior of [fac-Re(bpy)(CO)3Br] (bpy = 2,2'-bipyridine). Selective hydrogen evolution in ether solvents in the presence of triethylamine
- L27 ANSWER 27 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Air contaminants
- L27 ANSWER 28 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Toxic air pollutant emission factors a compilation for selected air toxic compounds and sources
- L27 ANSWER 29 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Advanced chemical fixation of organic and inorganic content wastes
- L27 ANSWER 30 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Toxic air pollutant/source crosswalk a screening tool for locating possible sources emitting toxic air pollutants
- L27 ANSWER 31 OF 48 ZCA COPYRIGHT 2005 ACS on STN

- TI Photo- and electrochemical reduction of carbon dioxide
- L27 ANSWER 32 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Steric effects on the solution chemistry of nickel(II) complexes with N-monomethylated 14-membered tetraaza macrocycles. The blue-to-yellow conversion and the oxidation and reduction behavior
- L27 ANSWER 33 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Electroplating in nonaqueous baths
- L27 ANSWER 34 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Enhanced photoredox chemistry in quantized semiconductor colloids
- L27 ANSWER 35 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Homogeneous catalysis of the photoreduction of water. 6. Mediation by polypyridine complexes of ruthenium(II) and cobalt(II) in alkaline media
- L27 ANSWER 36 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Electron transfer from .alpha.-aminoalkyl radicals to methylviologen
- L27 ANSWER 37 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Chemical oxidizability of organic components in water
- L27 ANSWER 38 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Photochemical hydrogen manufacture
- L27 ANSWER 39 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Tylosin derivatives and their pharmaceutical compositions
- L27 ANSWER 40 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Results of toxic action of water pollutants on Daphnia magna Straus tested by an improved standardized procedure
- L27 ANSWER 41 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Photochemical generation of carbon monoxide and hydrogen by reduction of carbon dioxide and water under visible light irradiation [artificial photosynthesis/solar-energy conversion/metal complex catalysis/tris(2,2'-bipyridine)ruthenium(II)/cobalt(II) system]
- L27 ANSWER 42 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Ethanol from methanol, carbon monoxide and hydrogen
- L27 ANSWER 43 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI 3-Amino-1-adamantylpenicillin
- L27 ANSWER 44 OF 48 ZCA COPYRIGHT 2005 ACS on STN

- TI 16-Substituted androstanes and androstenes
- L27 ANSWER 45 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Dihydrofusidic acid
- L27 ANSWER 46 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Studies on thiophene. XII
- L27 ANSWER 47 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Synthesis of thyroxine and related compounds. XVII.Preparation of some additional compounds related to thyroxine
- L27 ANSWER 48 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- TI Reserpine analogs. II. .beta.-Phenylethylamine derivatives
- => d 127 33 cbib abs hitstr hitrn
- L27 ANSWER 33 OF 48 ZCA COPYRIGHT 2005 ACS on STN
- 105:31929 Electroplating in nonaqueous baths. Kurosawa, Fumio (Nippon Steel Corp., Japan). Jpn. Kokai Tokkyo Koho JP 60245797 A2 19851205 Showa, 4 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 1984-100040 19840518.
- AB An electroplating process is carried out in a nonaq. bath consisting of metal complexes .gtoreq.0.01, solute dissolving in nonaq. solvent 0.05-10%, and nonaq. solvent. The method enables electroplating of Ti, Zr, and noble metals. Thus, Ti was electroplated at +500 mV in a MeOH soln. contg. 10% maleic anhydride and 1% LiCl on a C steel. The metal complex is a Ti-maleic anhydride complex.
- IT **75-05-8**, uses and miscellaneous

(electroplating from baths contg.)

- RN 75-05-8 ZCA
- CN Acetonitrile (8CI, 9CI) (CA INDEX NAME)

 $H_3C-C=N$

IT 102-71-6D, metal complexes

(electroplating from nonaq. bath contq.)

- RN 102-71-6 ZCA
- CN Ethanol, 2,2',2''-nitrilotris- (9CI) (CA INDEX NAME)

 $HO-CH_2-CH_2-N-CH_2-CH_2-OH$

ΙT	7440-02-0, uses and miscellaneous
	(electroplating of, from nonaq. bath)
RN	7440-02-0 ZCA
CN	Nickel (8CI, 9CI) (CA INDEX NAME)
Ni	
ΙΤ	75-05-8 , uses and miscellaneous (electroplating from baths contg.)
IT	102-71-6D, metal complexes
	(electroplating from nonaq. bath contg.
IT	7440-02-0, uses and miscellaneous
	(electroplating of, from nonaq. bath)